

# Career Review: Prof. Maurice Brookhart

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A selection of nearly 280 papers

Dong group at UT Austin  
Fanyang Mo  
April 24, 2013

## Maurice Brookhart (Organometallic Chemist)

William R. Kenan, Jr. Professor of Chemistry  
Department of Chemistry  
University of North Carolina at Chapel Hill (1969-present)

Born: 1943, North Carolina

### Education:

Johns Hopkins University (B. S., 1964)  
University of California, Los Angeles (Ph. D, 1968)  
Southampton University, England (Postdoc, 1969)

Publications: 274 papers (1964-present)

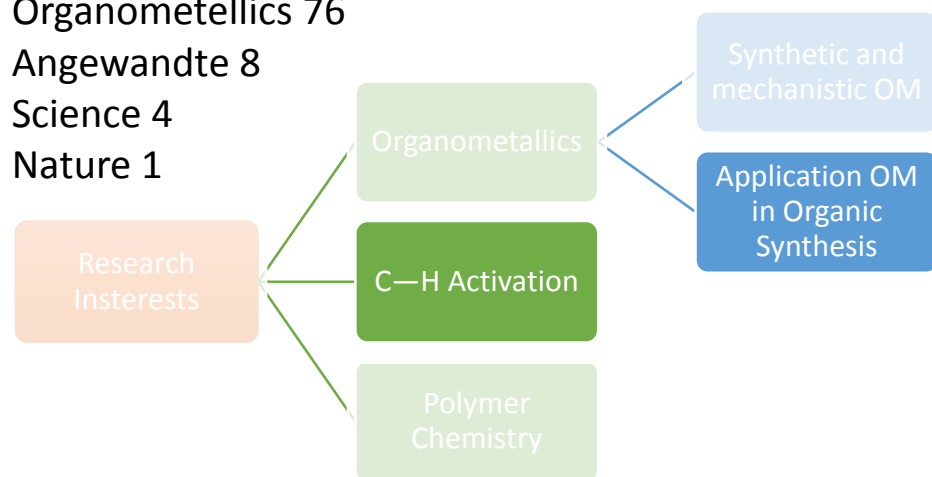
JACS 106

Organometallics 76

Angewandte 8

Science 4

Nature 1



III	IV	V	VI	VII	VIII	VIII	VIII	I	II
Sc 21 44.955	Ti 22 47.882	V 23 50.941	Cr 24 51.995	Mn 25 54.938	Fe 26 55.845	Co 27 58.933	Ni 28 58.693	Cu 29 63.546	Zn 30 65.38
Y 39 88.905	Zr 40 91.224	Nb 41 92.906	Mo 42 95.94	Tc 43 98.906	Ru 44 101.07	Rh 45 102.905	Pd 46 106.42	Ag 47 107.868	Cd 48 112.411
*	Hf 72 178.49	Ta 73 180.948	W 74 183.84	Re 75 186.207	Os 76 190.23	Ir 77 192.222	Pt 78 195.084	Au 79 196.967	Hg 80 200.59
**	Rf 104 261	Db 105 262	Sg 106 263	Bh 107 264	Hs 108 265	Mt 109 266	Ds 110 267	Rg 111 268	Cn 112 269
3	4	5	6	7	8	9	10	11	12

Latest 15 years (Since 1998)

# Contents

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## 1. Inert Bond Activation (Co, Rh)

- C—H Activation
- C—C Activation

## 2. Dehydrogenation (Ir)

## 3. Reduction of Alkyl Halides and Carbonyl compounds (Ir)

## 4. Miscellaneous

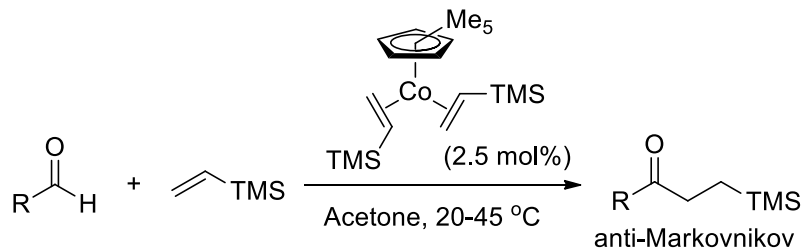
## 5. Acknowledgement

VIII

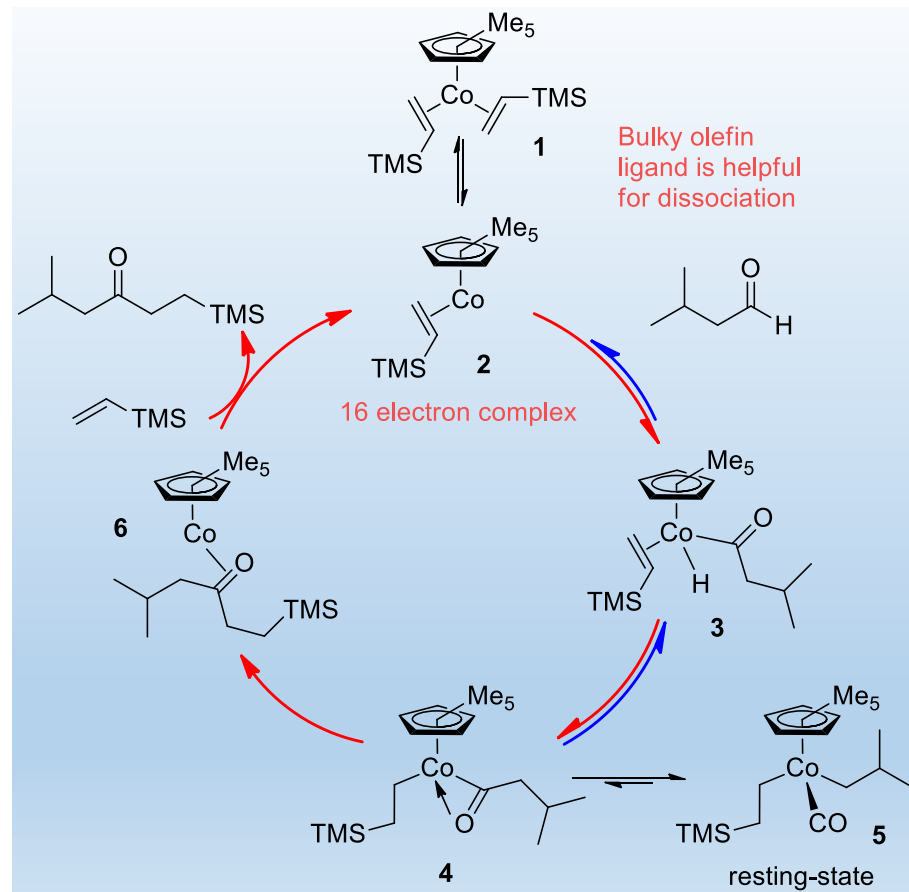
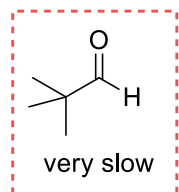
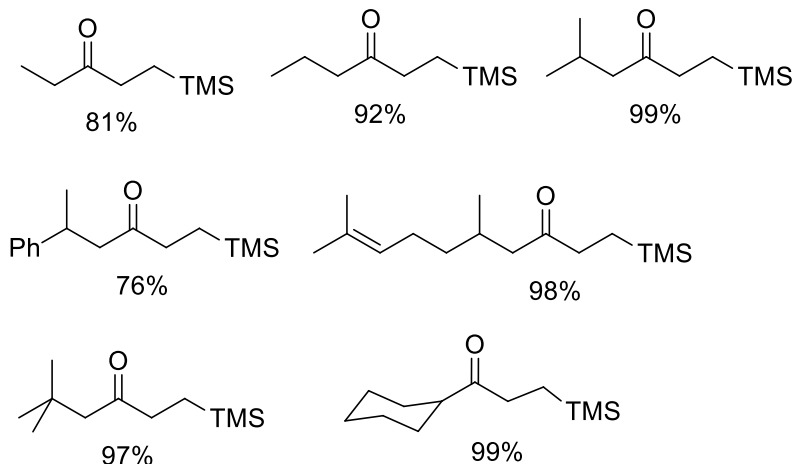
Cobalt	27
Co	58.933
Rhodium	45
Rh	102.91
Iridium	77
Ir	192.22

# 1. Inert Bond Activation *C—H* bond

## 1) Co(I) catalyzed addition of Aliphatic Aldehyde to Vinyl Silanes

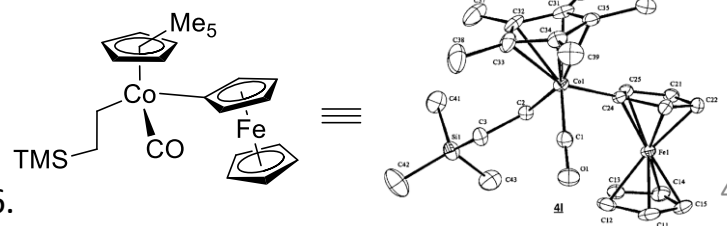


Selected examples:



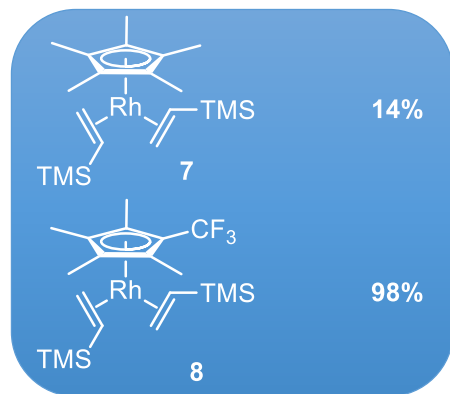
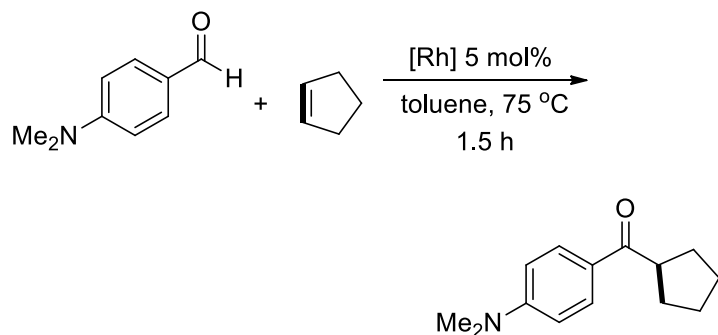
Lenges, C. P., et. al. *J. Am. Chem. Soc.* **1998**, *120*, 6965-6979.

Aromatic version, see: *J. Am. Chem. Soc.* **1997**, *119*, 3165-3166.

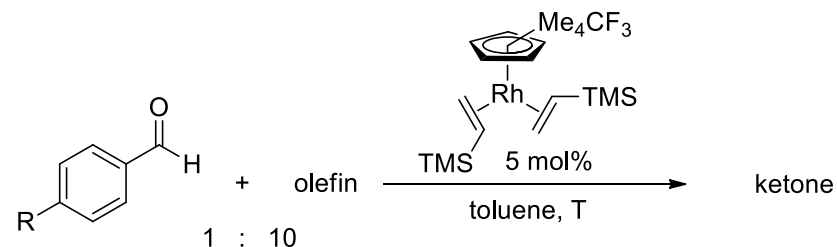


# 1. Inert Bond Activation *C—H bond*

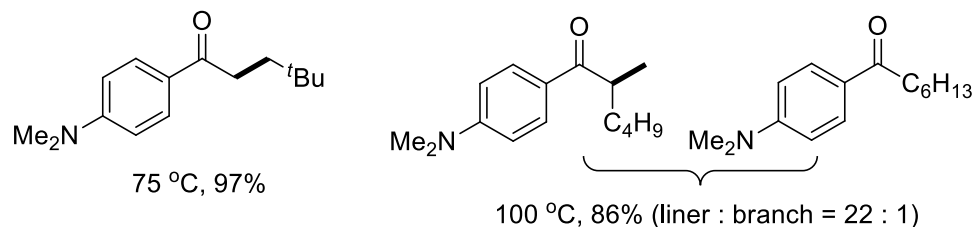
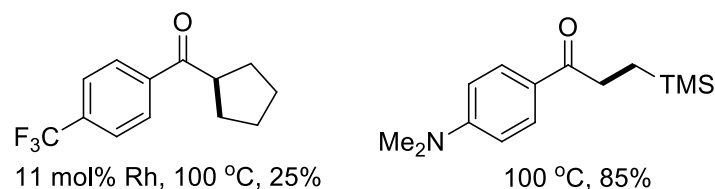
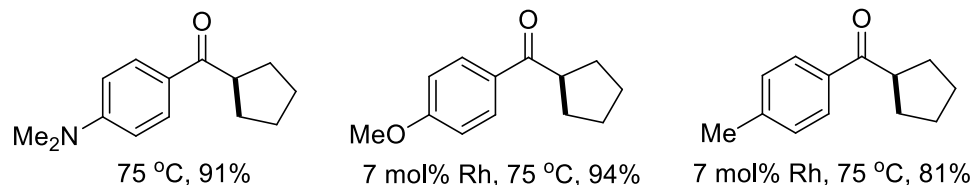
## 2) Rh(I) catalyzed addition of Aromatic Aldehyde to Olefins



\*Reductive elimination would increase for a more electron deficient metal center.



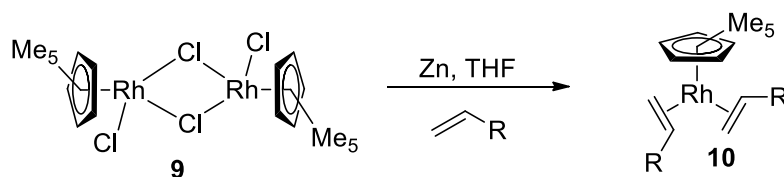
Selected examples:



# 1. Inert Bond Activation *C—H* bond

## 3) [C<sub>5</sub>Me<sub>5</sub>Rh(olefin)<sub>2</sub>] catalyzed Hydrogen/Deuterium Exchange Reactions

Synthesis of [C<sub>5</sub>Me<sub>5</sub>Rh(olefin)<sub>2</sub>]:



**10a**: R = SiMe<sub>3</sub>

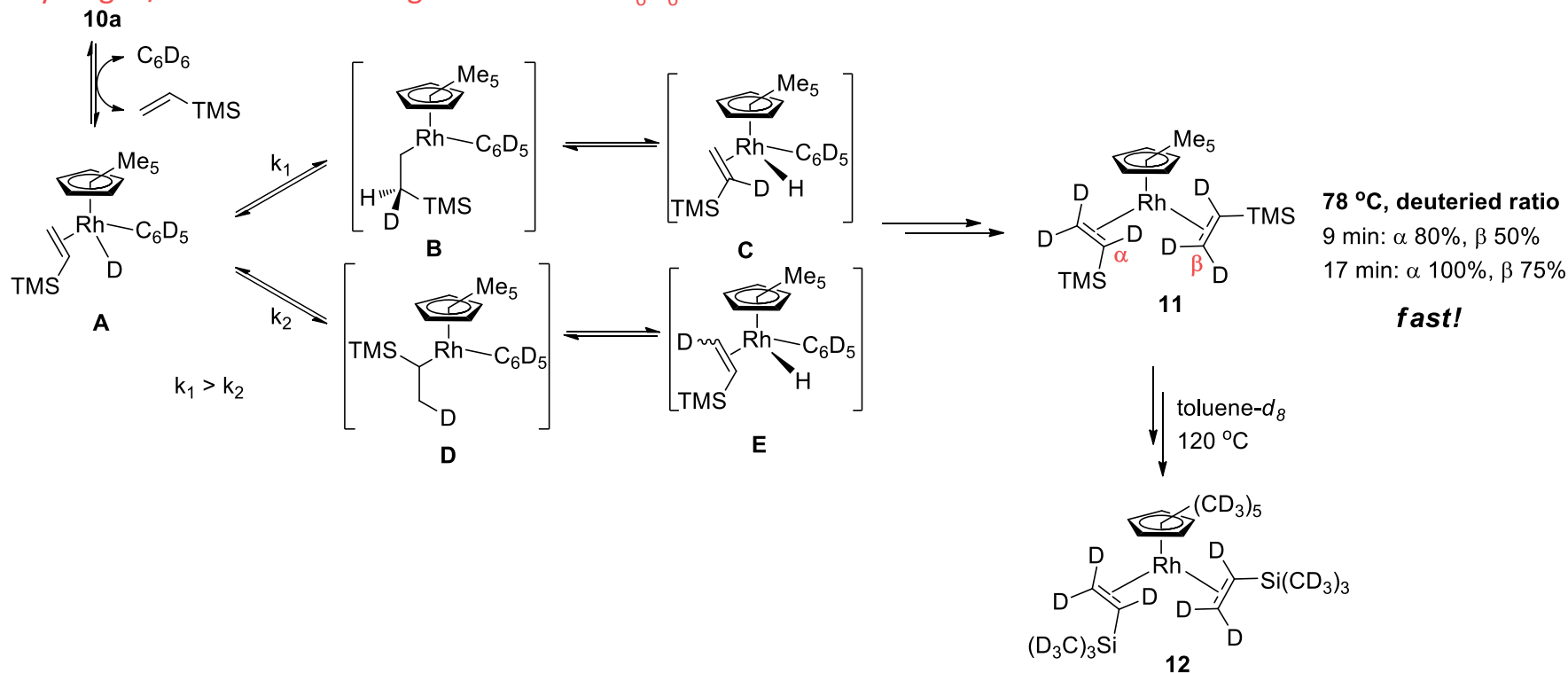
**10b**: R = SiMe<sub>2</sub>OEt

**10c**: R = Si(O<sup>i</sup>Pr)<sub>3</sub>

**10d**: R = SiMe(OSiMe<sub>3</sub>)<sub>2</sub>

**10e**: R = SiPh<sub>2</sub>O<sup>i</sup>Pr

Hydrogen/Deuterium Exchange Reactions in C<sub>6</sub>D<sub>6</sub>:



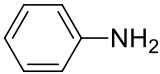
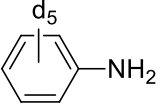
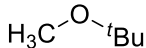
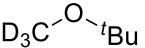
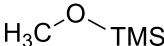
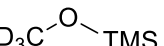
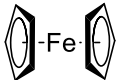
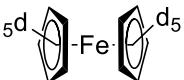

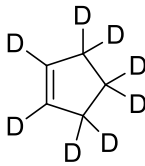
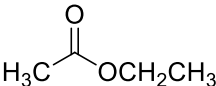
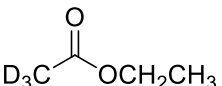
Lenges, C. P., et. al. *J. Am. Chem. Soc.* **1999**, *121*, 4385-4396;

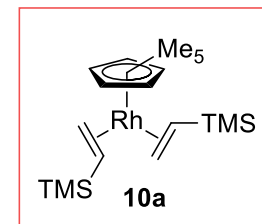
Seiwell, L. P. *J. Am. Chem. Soc.* **1974**, *96*, 7134-7135.

# 1. Inert Bond Activation *C—H bond*

## 3) [C<sub>5</sub>Me<sub>5</sub>Rh(olefin)<sub>2</sub>] catalyzed Hydrogen/Deuterium Exchange Reactions

**Table 1.** H/D Exchange between Benzene-*d*<sub>6</sub> to Other Substrates Catalyzed by **10a**<sup>a</sup>

entry	substrate	product	%deuteration <sup>b</sup>	
			5h	24h
1			<i>ortho</i> : 76 <i>meta</i> : 94 <i>para</i> : 93	91 97 97
2			31	44
3			46	62
4			93	-
5			49	46
6			61	80



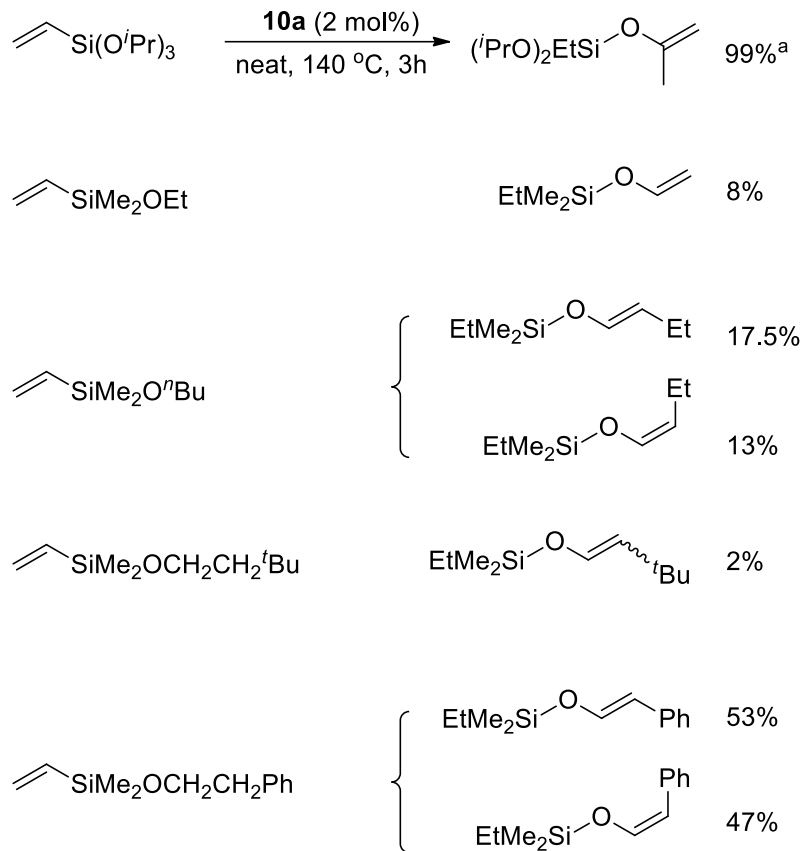
<sup>a</sup> **8a** (0.01 g, 2.3×10<sup>-5</sup> mol), substrate (4.6×10<sup>-4</sup> mol) in C<sub>6</sub>D<sub>6</sub> at 110 °C.

<sup>b</sup> %deuteration estimated from residual <sup>1</sup>H NMR signal intensities.

# 1. Inert Bond Activation *C—H* bond

## 3) [C<sub>5</sub>Me<sub>5</sub>Rh(olefin)<sub>2</sub>] catalyzed Hydrogen/Deuterium Exchange Reactions

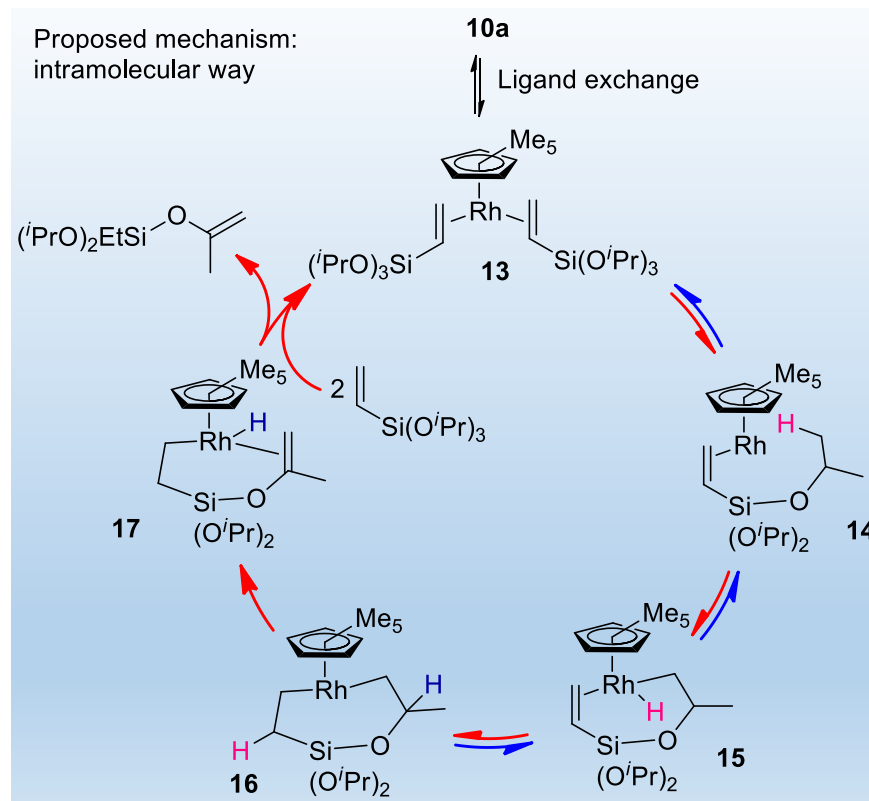
**Table 2.** Isomerization of Alkoxysilanes to Silyl Enolates



<sup>a</sup>NMR yield.

Lenges, C. P., *et. al. J. Am. Chem. Soc.* **1999**, *121*, 4385-4396.

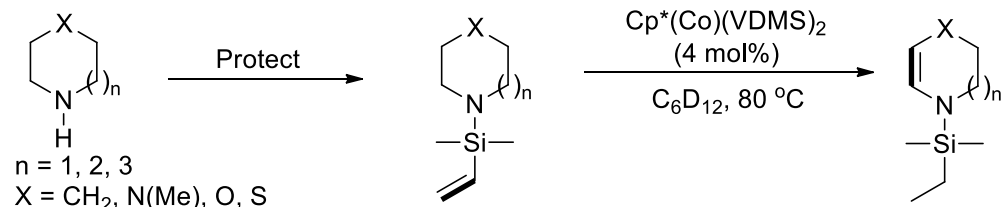
Diaz-Requejo, M. N., *et. al. J. Am. Chem. Soc.* **2003**, *125*, 2038-2039.





# 1. Inert Bond Activation *C—H* bond

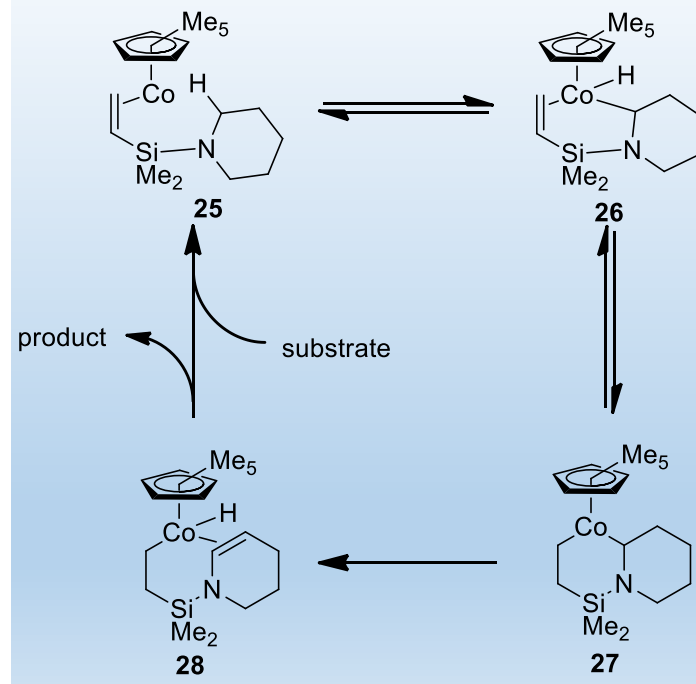
## 4) Co(I) Catalyzed $sp^3$ C—H Bond: Catalytic Synthesis of Enamines



Substrate	Product	Yield
		80 °C, 6h, >99%
		80 °C, 6h, >99%
		80 °C, 2h, >95%
		80 °C, 2h, >99%
		M = Co, 80 °C, 2h, 13% M = Rh, 140 °C, 1h, >90%
		80 °C, 6d, >90%
		80 °C, 6d, >95%

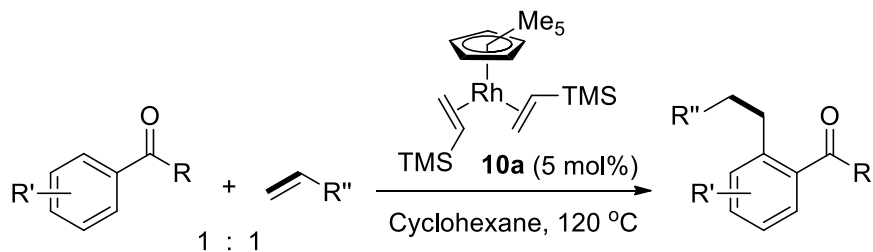
VDMS: vinyl(dimethyl)silyl; EDMS: ethyl(dimethyl)silyl.

Proposed mechanism:

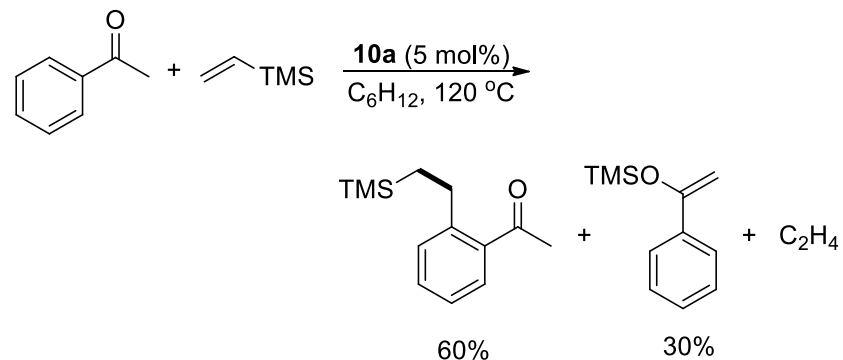
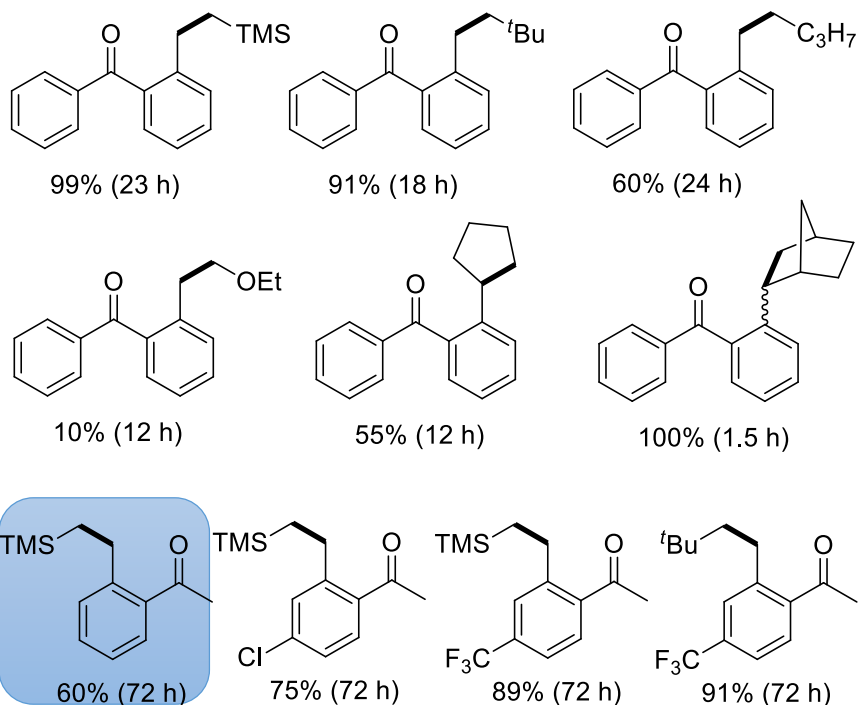


# 1. Inert Bond Activation *C—H* bond

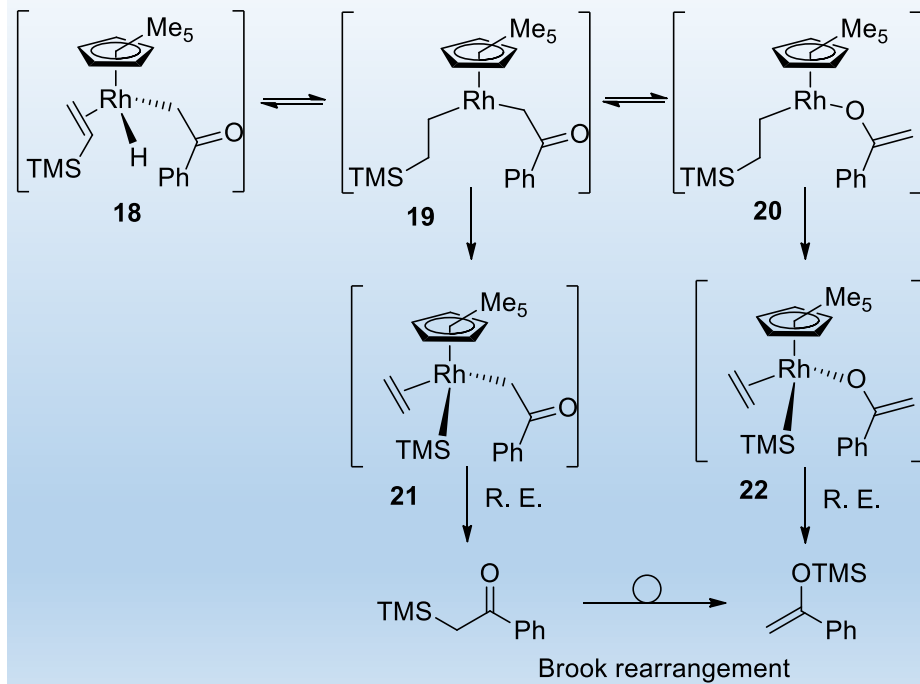
## 5) [C<sub>5</sub>Me<sub>5</sub>Rh(olefin)<sub>2</sub>] catalyzed Addition of Olefins to Aromatic Ketones



Selected examples:

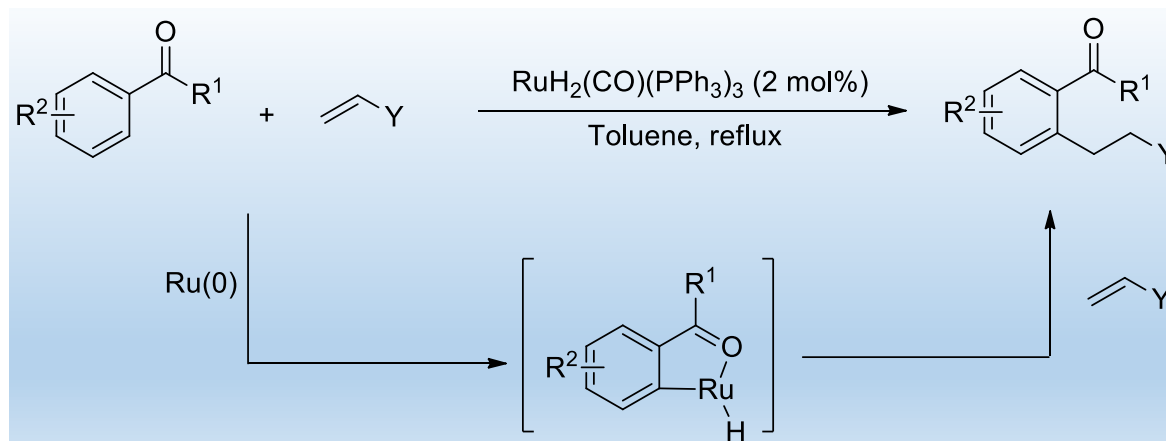
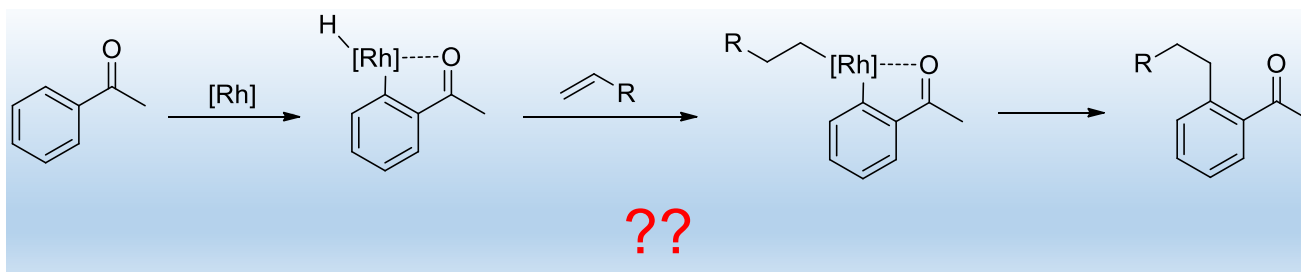


Possible pathway:



# 1. Inert Bond Activation *C—H bond*

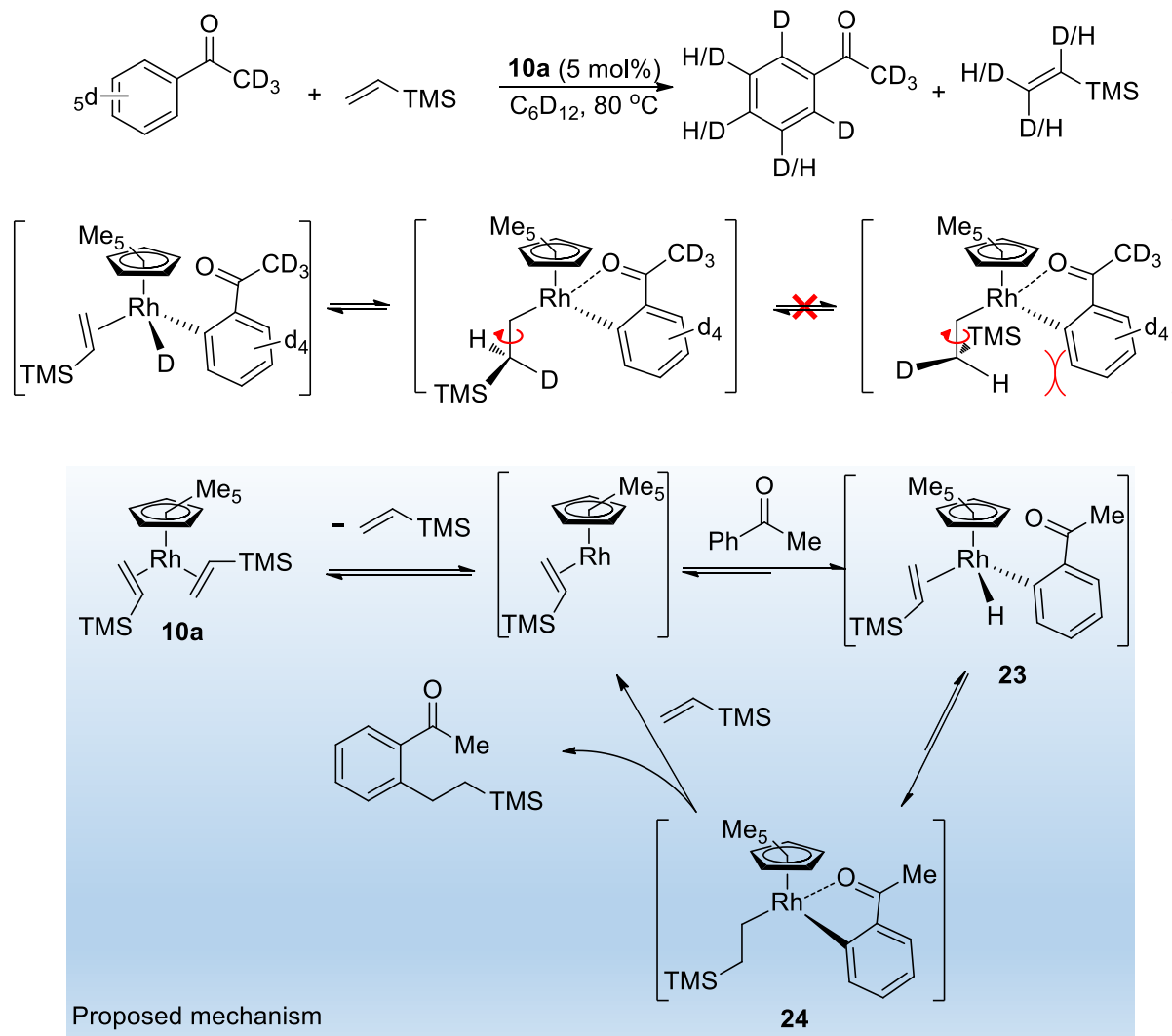
## 5) $[\text{C}_5\text{Me}_5\text{Rh}(\text{olefin})_2]$ catalyzed Addition of Olefins to Aromatic Ketones



Murai, S., *et. al. Nature* **1993**, 366, 529-531.

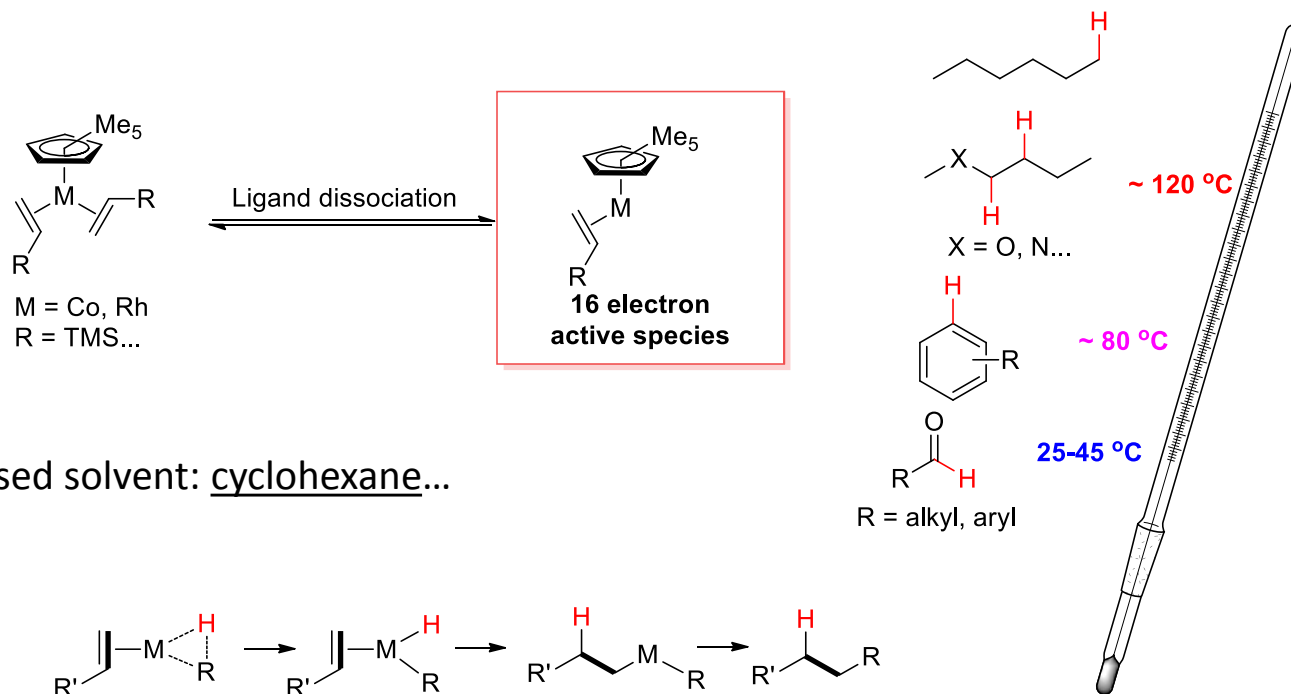
# 1. Inert Bond Activation *C—H bond*

## 5) $[C_5Me_5Rh(olefin)_2]$ catalyzed Addition of Olefins to Aromatic Ketones



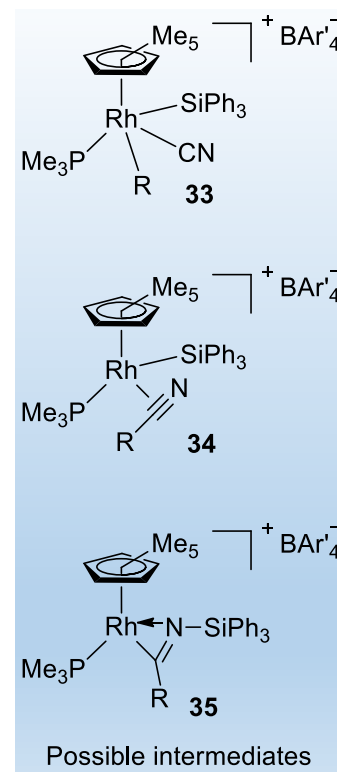
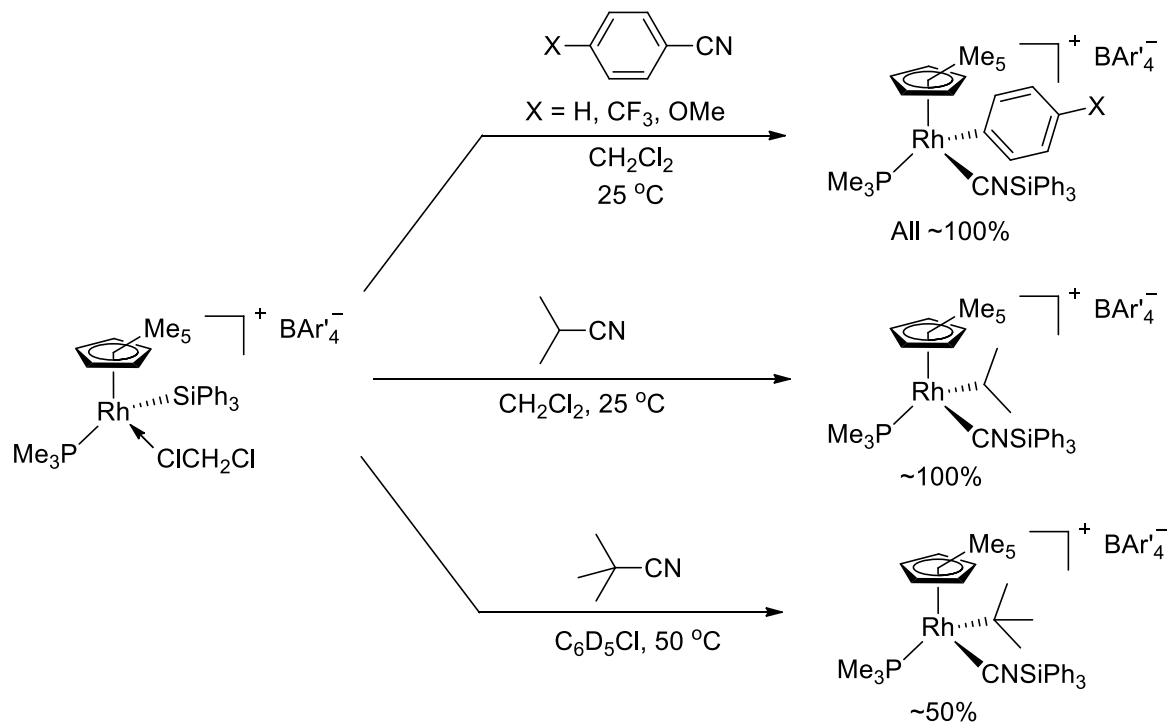
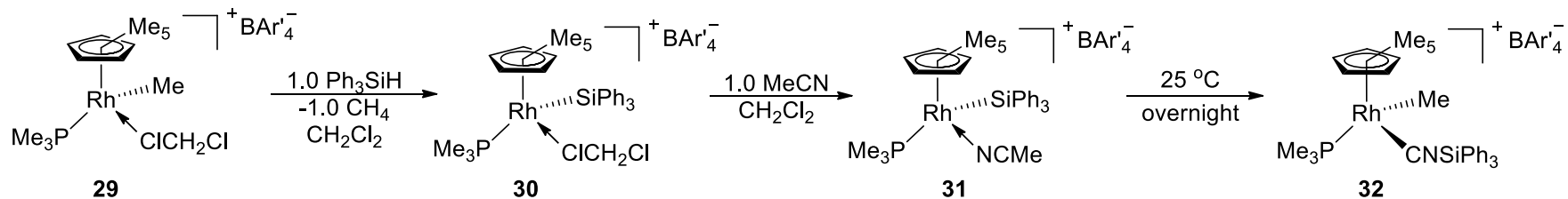
# 1. Inert Bond Activation *C—H* bond

## 6) Summary for Cp\**M*(olefin)<sub>2</sub> catalyzed C—H bond activation.



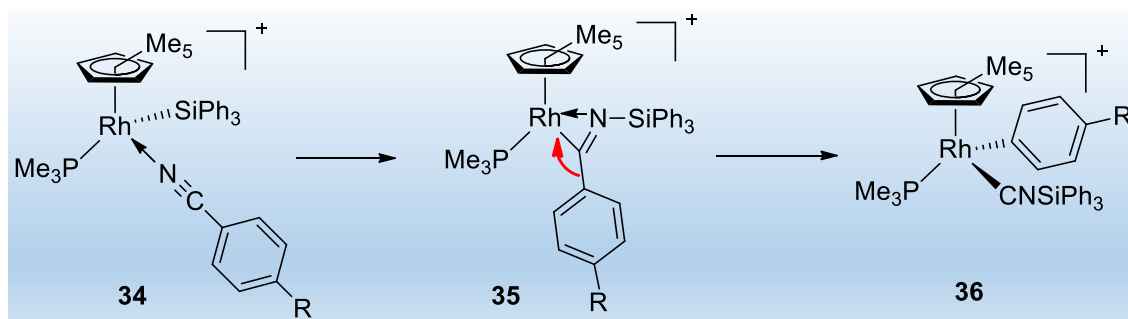
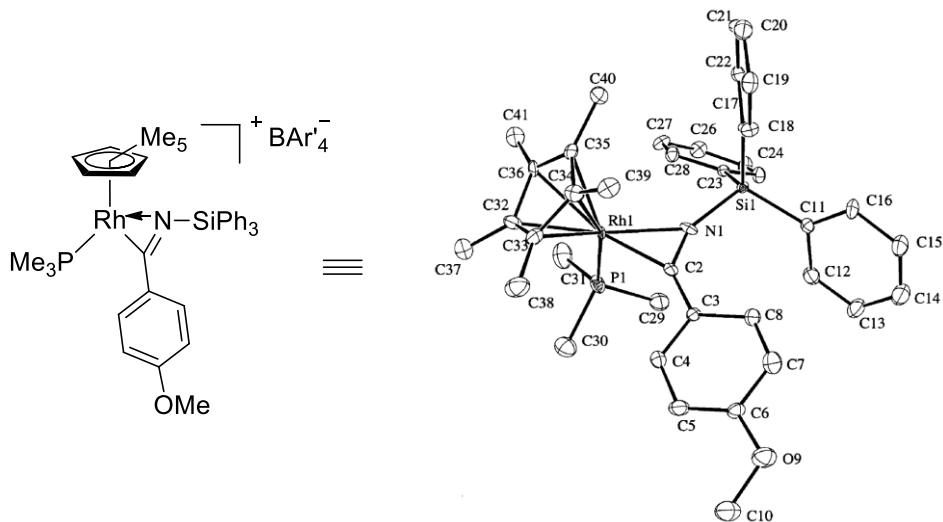
# 1. Inert Bond Activation *C—C bond*

## 8) C—C bond activation of R—CN Using a Cationic Rh(III) complex.



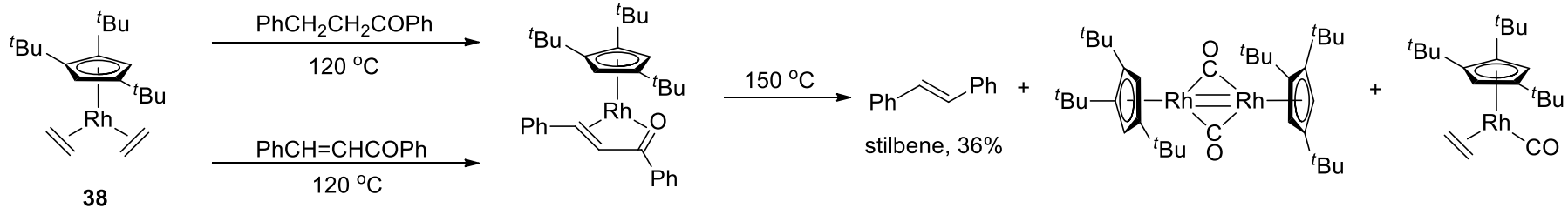
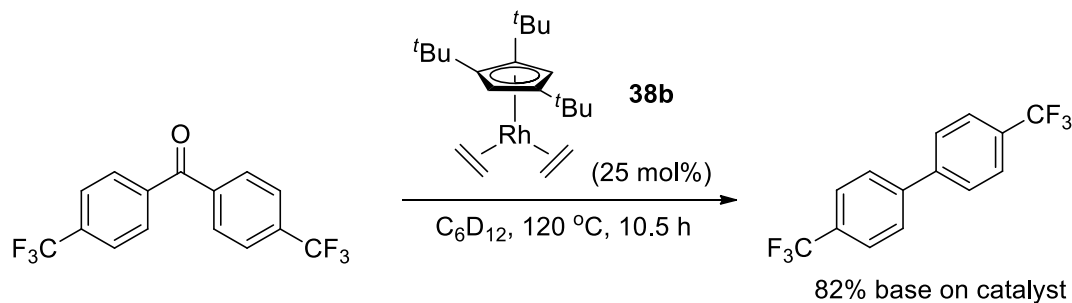
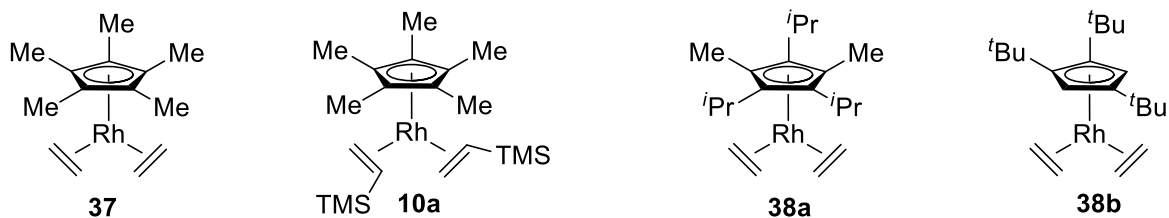
# 1. Inert Bond Activation *C—C bond*

8) C—C bond activation of R—CN Using a Cationic Rh(III) complex.



# 1. Inert Bond Activation *C—C bond*

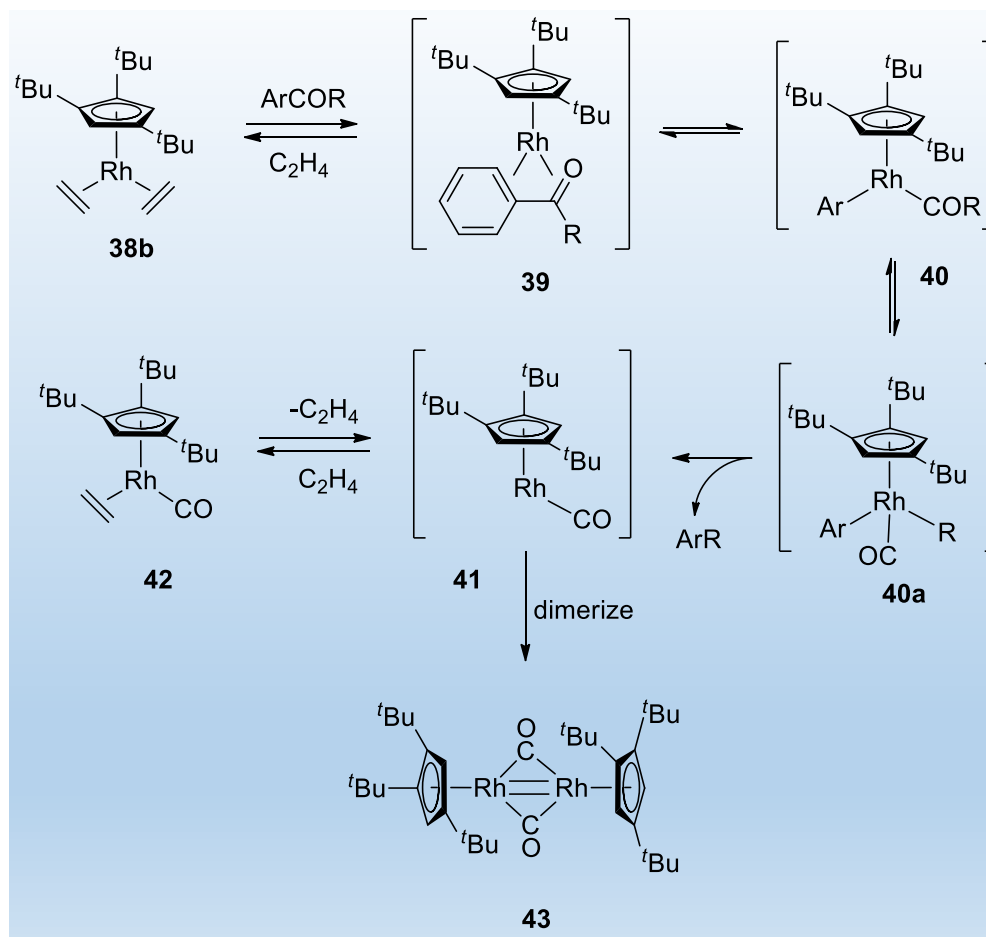
## 9) Decarbonylation of Aryl Ketones Mediated by Bulky Cp\*Rh(ethylene)<sub>2</sub>.





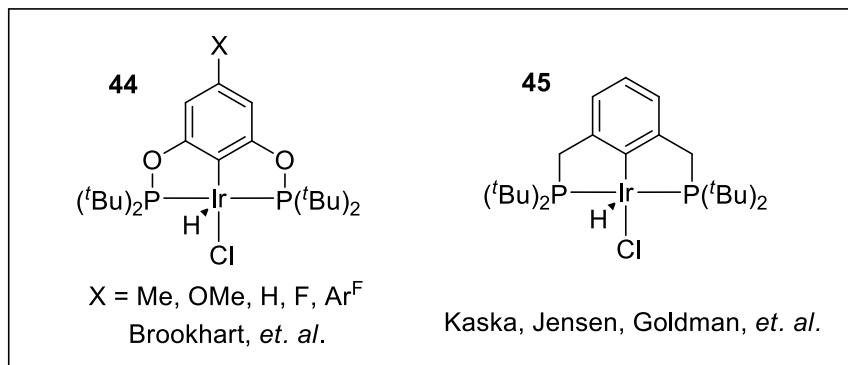
# 1. Inert Bond Activation *C—C bond*

## 9) Decarbonylation of Aryl Ketones Mediated by Bulky Cp\*Rh(ethylene)<sub>2</sub>.

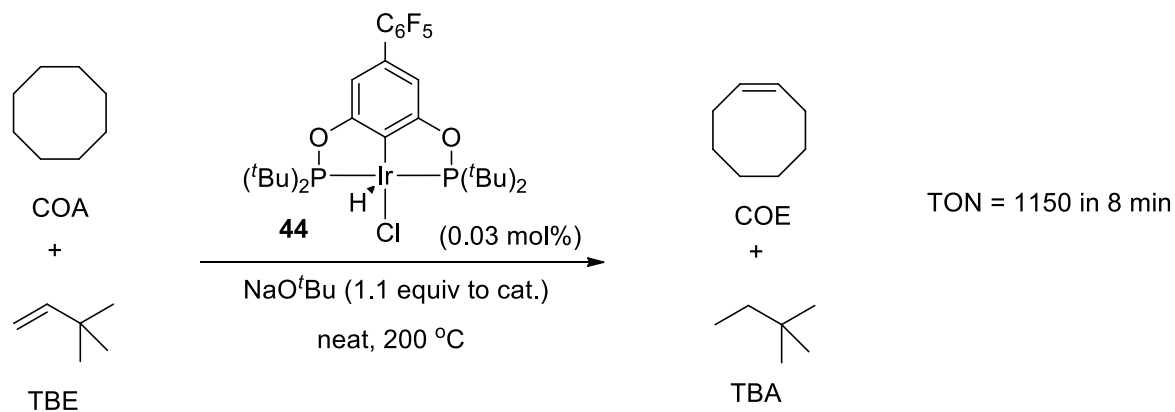


## 2. Dehydrogenation

1) Ir Pincer complex catalyzed alkane dehydrogenation reaction.



Catalyst for alkane dehydrogenation



Göttker-Schnetmann, I., *et. al. J. Am. Chem. Soc.* **2004**, *126*, 1804-1811.

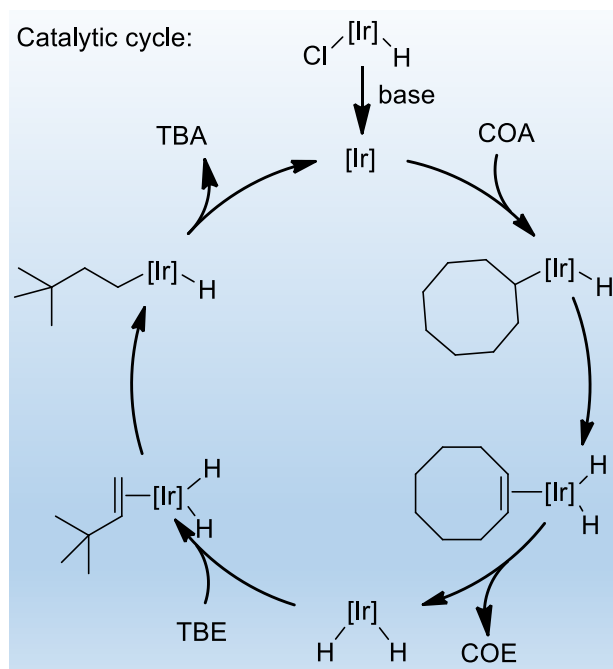
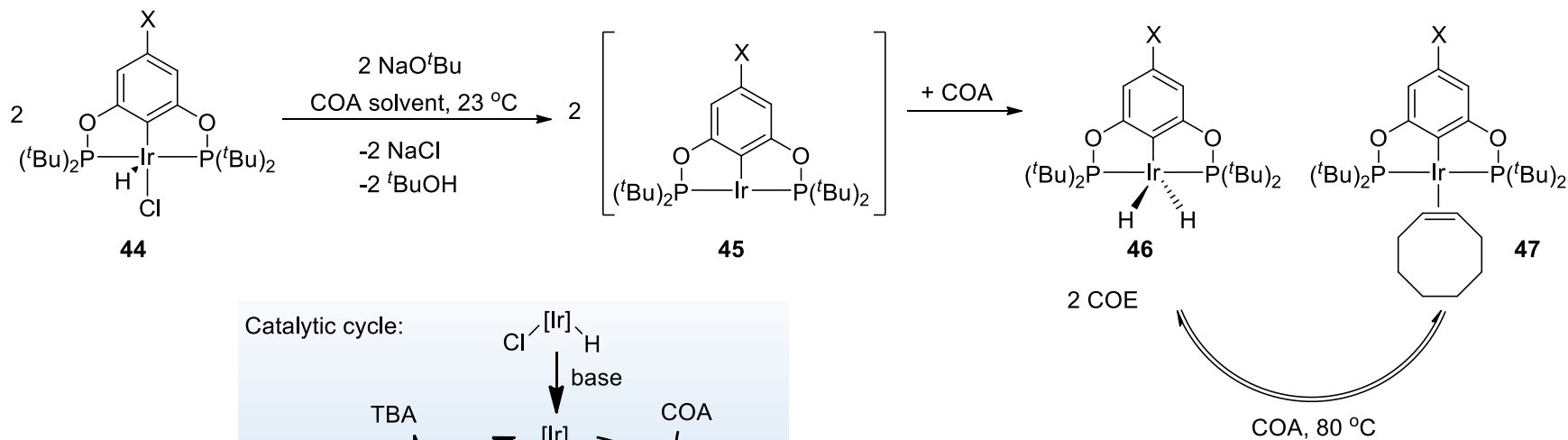
Göttker-Schnetmann, I., *et. al. Organometallics* **2004**, *23*, 1766-1776.

For review, see: Goldman, A. S., *et. al. Chem. Rev.* **2011**, *111*, 1761-1779.

## 2. Dehydrogenation

1) Ir Pincer complex catalyzed alkane dehydrogenation reaction.

Mechanism Study:



Other related papers:

*J. Am. Chem. Soc.* **2006**, *128*, 17114-17119.

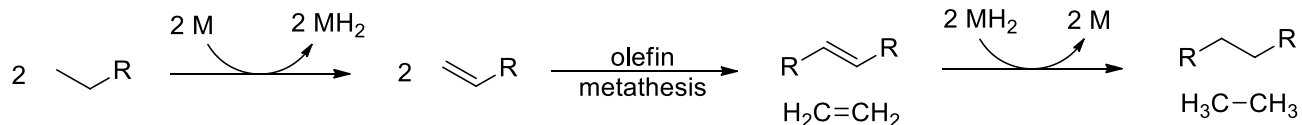
*Advanced Synthesis & Catalysis* **2009**, *351*, 188-206.

*Organometallics* **2009**, *28*, 5432-5444.

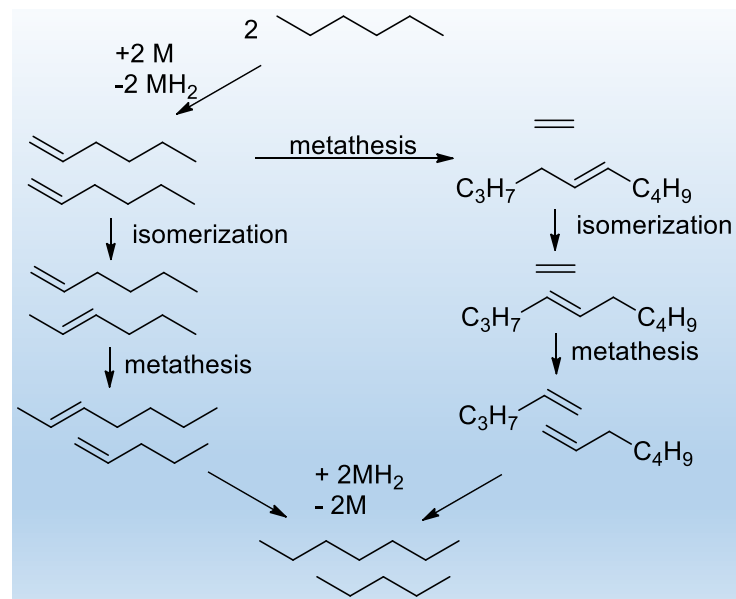
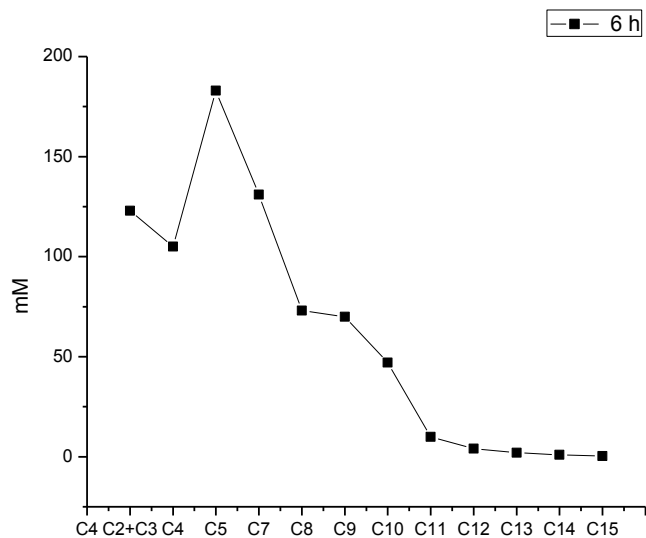
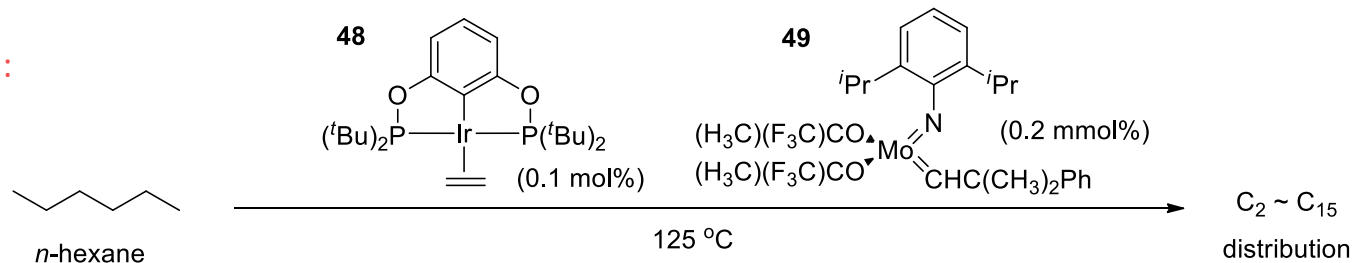
## 2. Dehydrogenation

### 2) Catalytic Alkane Metathesis by Tandem Alkane Dehydrogenation-Olefin Metathesis.

The idea:



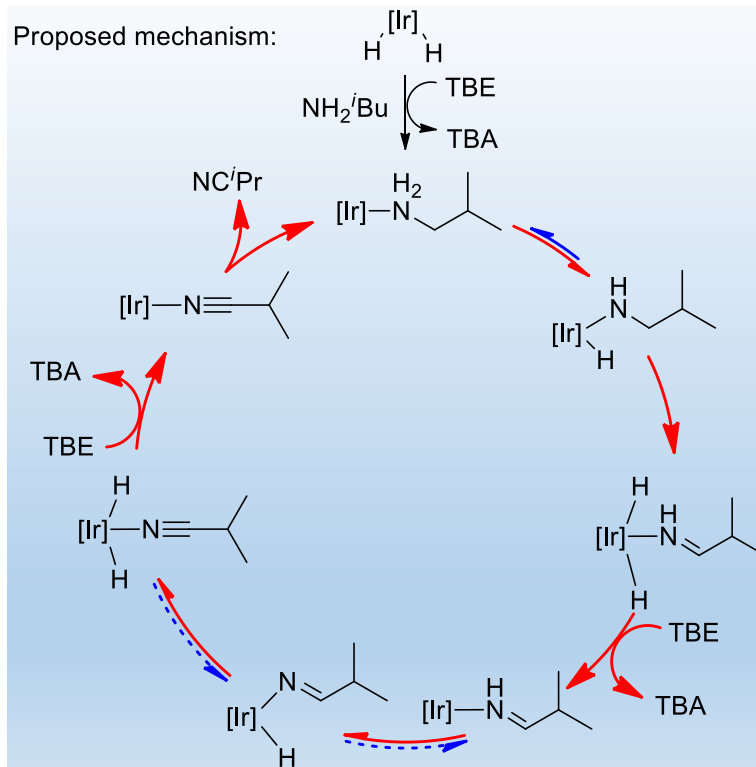
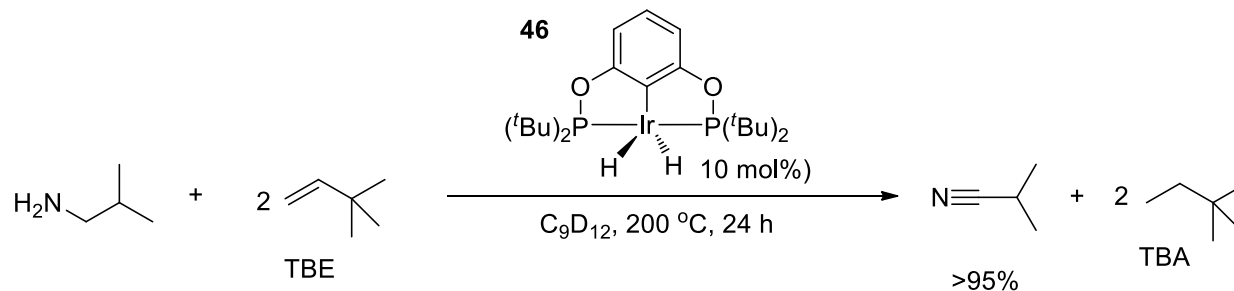
The reaction:



Goldman, A. S., *et al. Science* **2006**, 312, 257-261.

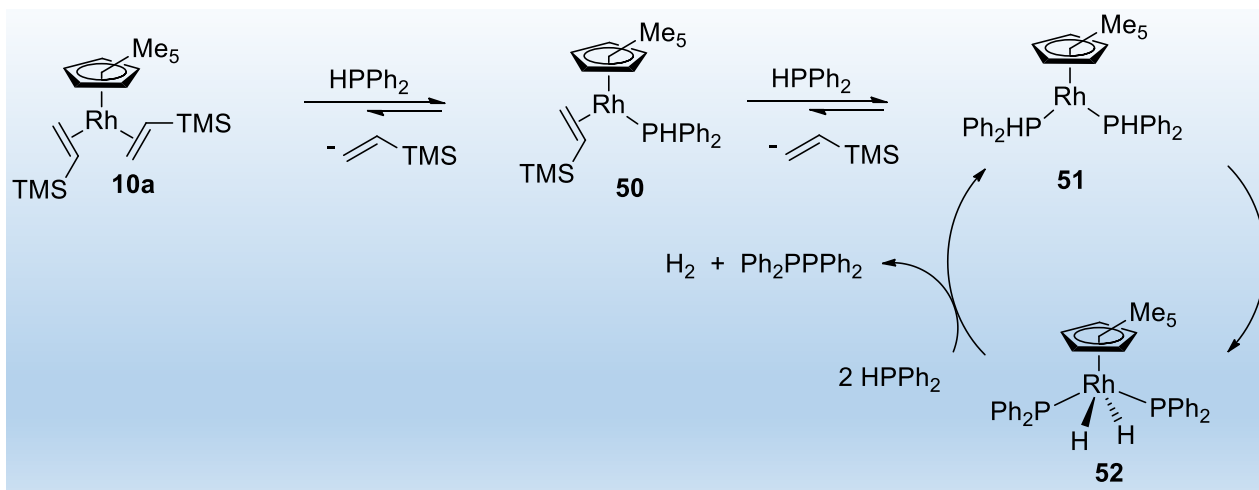
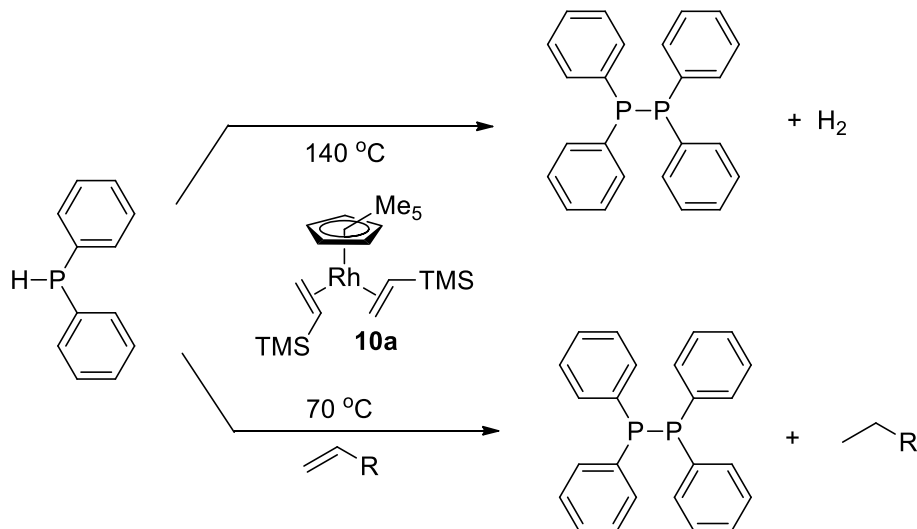
## 2. Dehydrogenation

### 3) Ir Pincer complex-Catalyzed Dehydrogenation of Primary Amines to Nitriles.



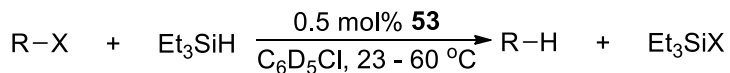
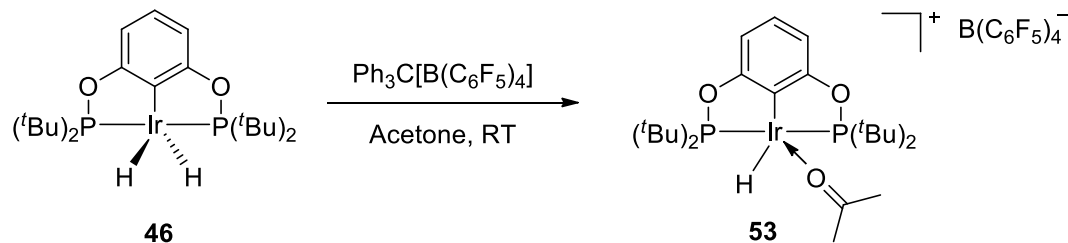
## 2. Dehydrogenation

### 4) Rh(I)-catalyzed Dihydrocoupling of Phosphanes.



### 3. Reduction reaction

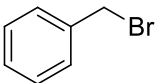
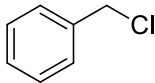
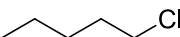
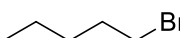
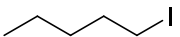
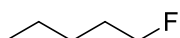
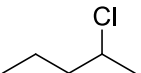
#### 1) Ir Pincer complex-Catalyzed Reduction of Alkyl Halides by Triethylsilane.



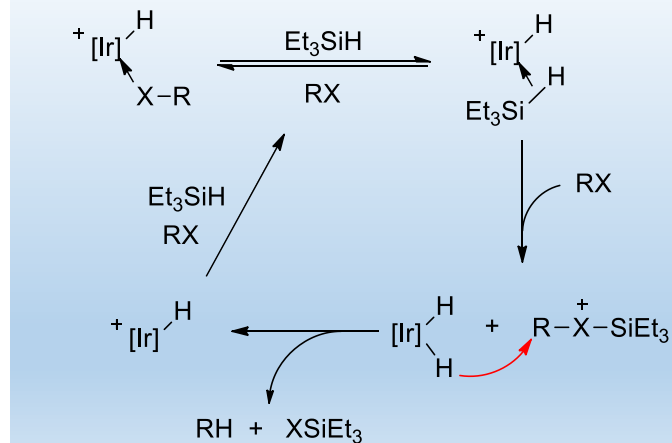
Reaction rate:

**Separate flasks** R-Br > R-Cl > R-I

**Same flask** R-I > R-Br > R-Cl

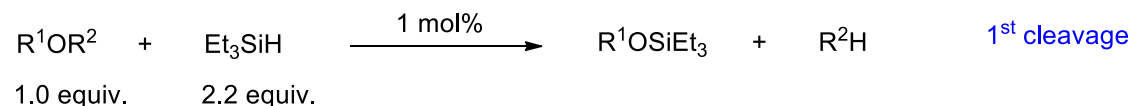
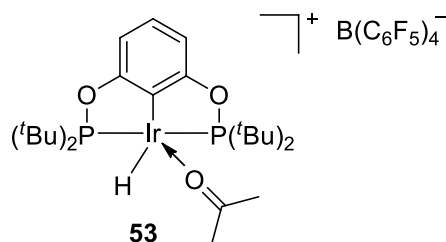
Halide	Yield
	23 °C, 0.3 h, 99%
	23 °C, 0.3 h, 99%
	60 °C, 7 h, 99%
	60 °C, 1.5 h, 99%
	60 °C, 48 h, 99%
	60 °C, 50 h, 92% (2 mol% <b>A</b> )
	60 °C, 2.3 h, 99%

Proposed mechanism:

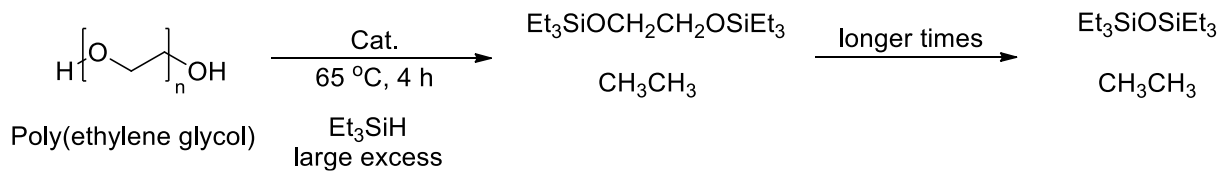
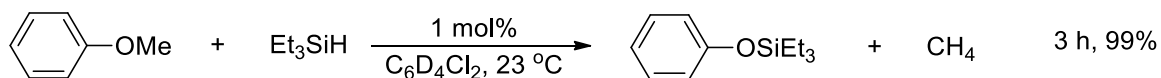
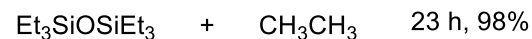
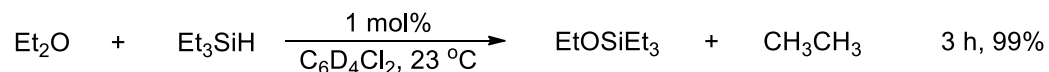


### 3. Reduction reaction

#### 2) Ir Pincer complex-Catalyzed Cleavage of Alkyl Ethers with Triethylsilane.



Four examples:

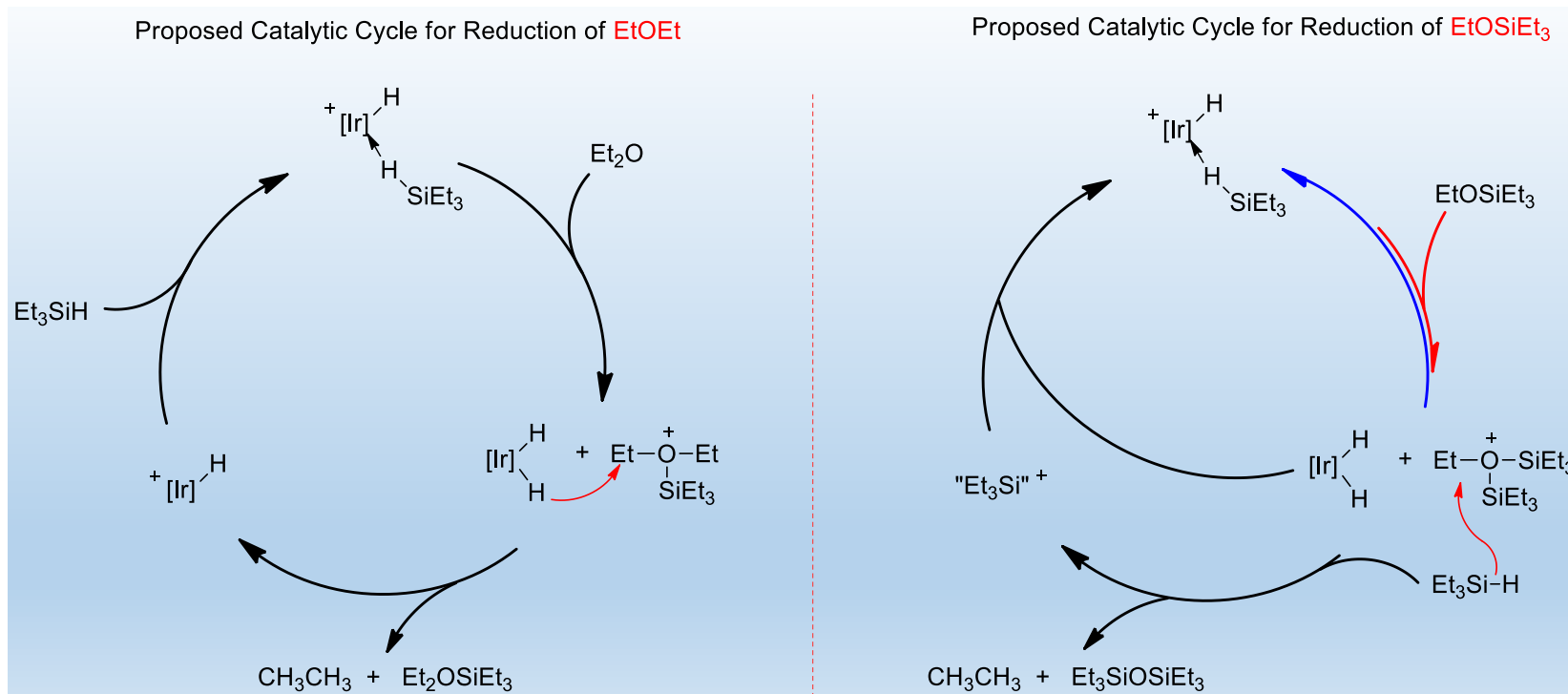


Limitation: Aromatic ether does not work



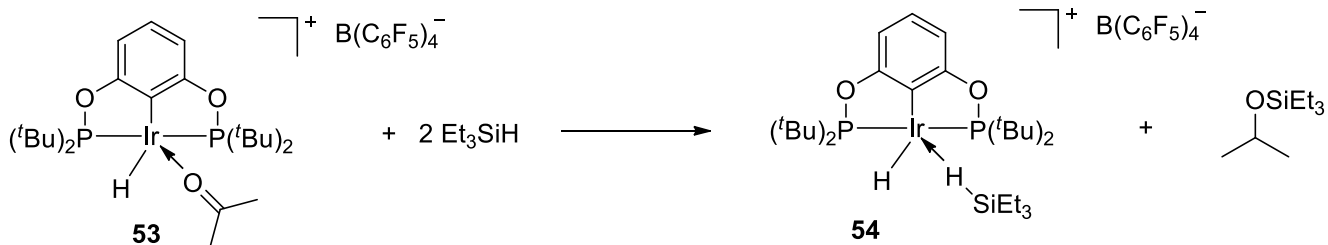
### 3. Reduction reaction

#### 2) Ir Pincer complex-Catalyzed Cleavage of Alkyl Ethers with Triethylsilane.



# 3. Reduction reaction

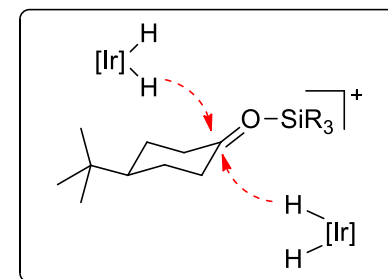
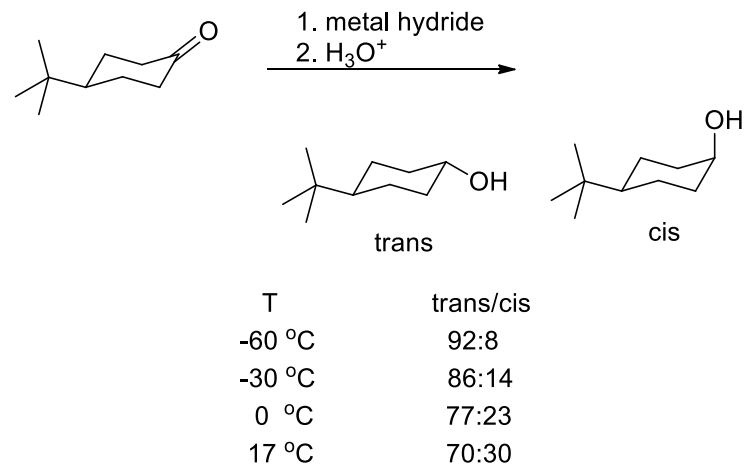
## 3) Ir Pincer complex-Catalyzed Hydrosilylation of Ketones and Aldehydes.



**Table 3.** Hydrosilylation of Carbonyl Functions

entry	substrate	t / h	conversion / %	product
1		0.3	quant.	
2		0.5	quant.	
3		0.3	quant.	
4		0.3	quant.	
5		0.3	quant.	$\text{EtOSiEt}_3$ (70%) + $\text{Et}_2\text{O}$ (30%)

$\xrightarrow{\text{1}^{\text{st}} \text{ hydrosilylation}}$

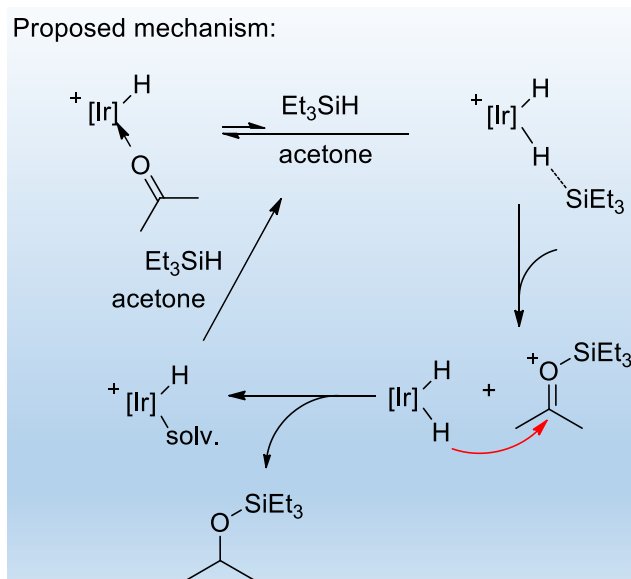


0.5 mol% cat., 3 equiv.  $\text{Et}_3\text{SiH}$ , RT,  $\text{C}_6\text{D}_5\text{Cl}$

Park, S., et al. *Organometallics* **2010**, *29*, 6057-6064.

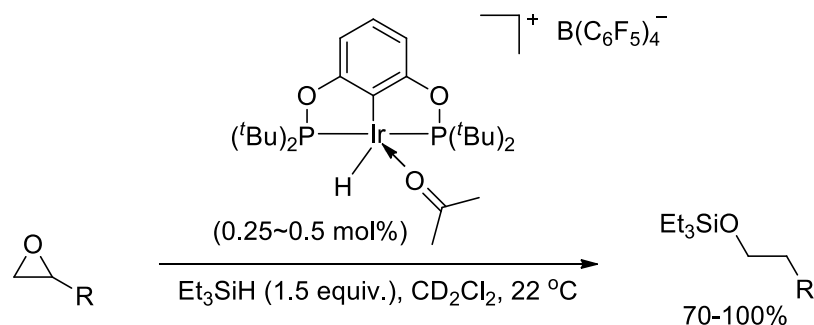
### 3. Reduction reaction

#### 3) Ir Pincer complex-Catalyzed Hydrosilation of Ketones and Aldehydes.



Park, S., et. al. *Organometallics* **2010**, 29, 6057-6064.

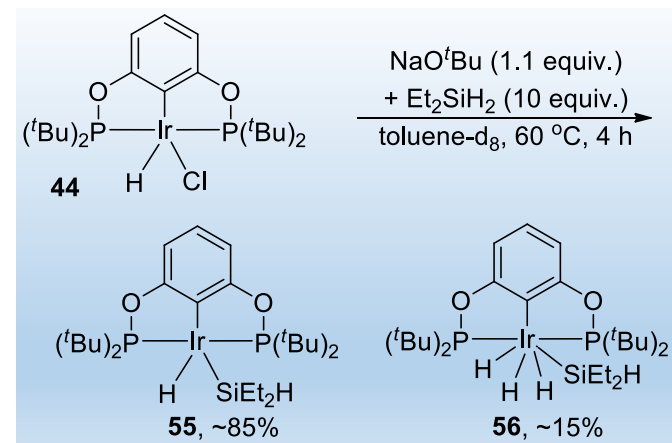
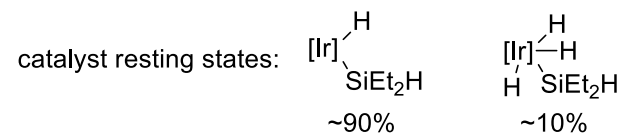
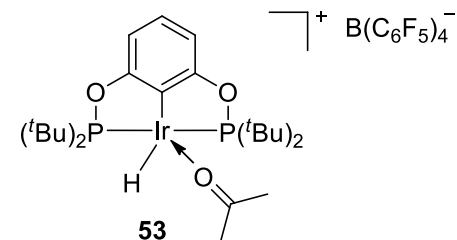
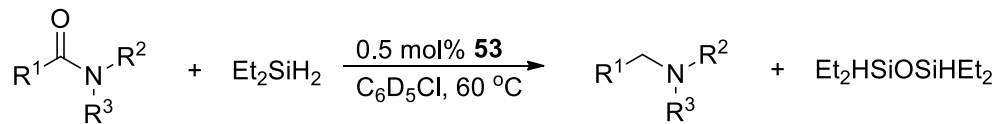
#### 4) Ir Pincer complex-Catalyzed Hydrosilation of Epoxides.



Park, S., et. al. *Chem. Commun.* **2011**, 47, 3643-3645.

### 3. Reduction reaction

#### 5) Ir Pincer complex-Catalyzed Reduction of Amides to Amines.



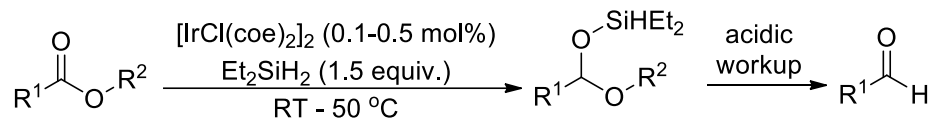
entry	substrate	t / h	product	yield
1		1.5 EtMe <sub>2</sub> SiH 22 h, 66%		94%
2		1		quant.
3		0.5		quant.
4		5		quant.
5		1		quant.
6		1.2		quant.

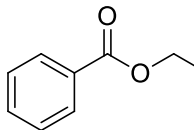
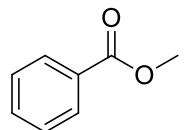
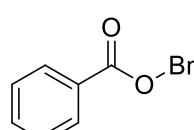
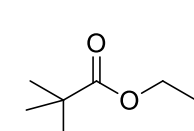
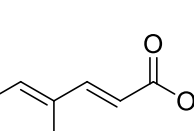
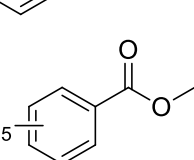
0.5 mol% cat, 3 equiv. of Et<sub>2</sub>SiH<sub>2</sub>

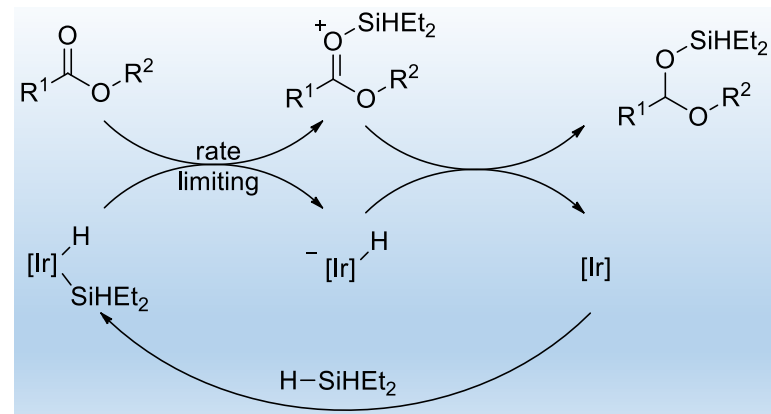
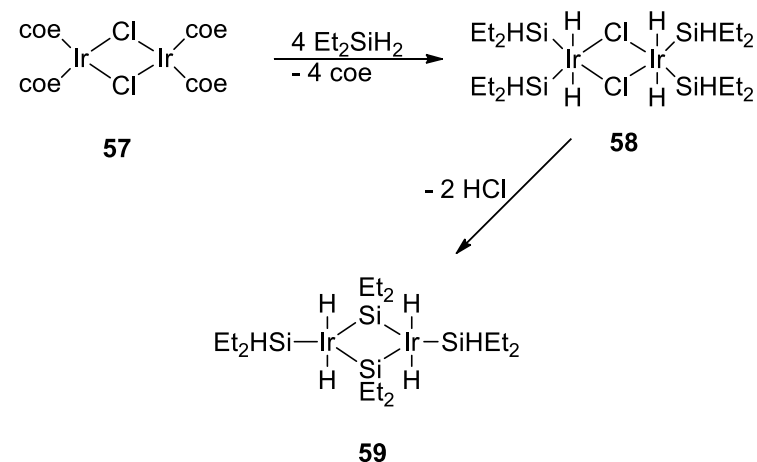
Park, S., *et. al. J. Am. Chem. Soc.* **2012**, *134*, 640-653.

## 3. Reduction reaction

### 6) Ir-Catalyzed Reduction of Esters to Aldehydes.

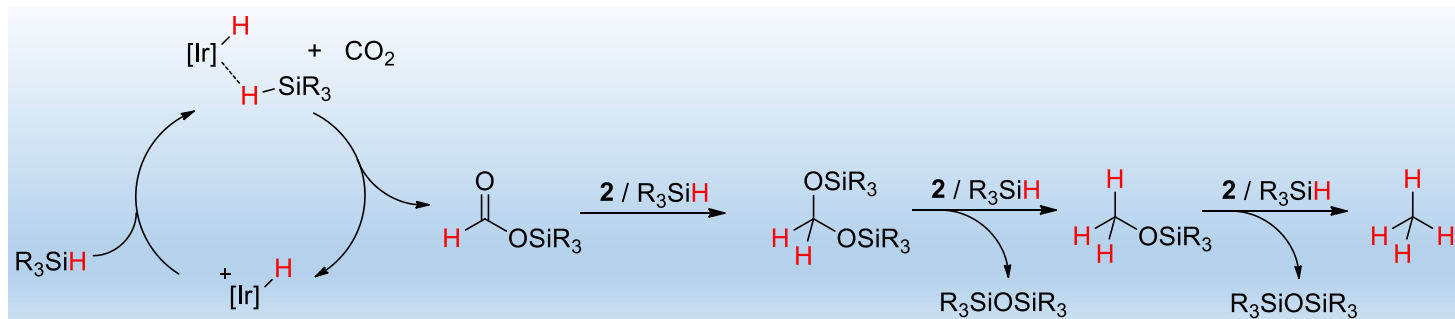


entry	substrate	T / °C	t / h	yield
1		23	1	95%
2		23	2	99%
3		23	1	99%
4		23	3	99%
5		23	1	99%
6		50	4	90%



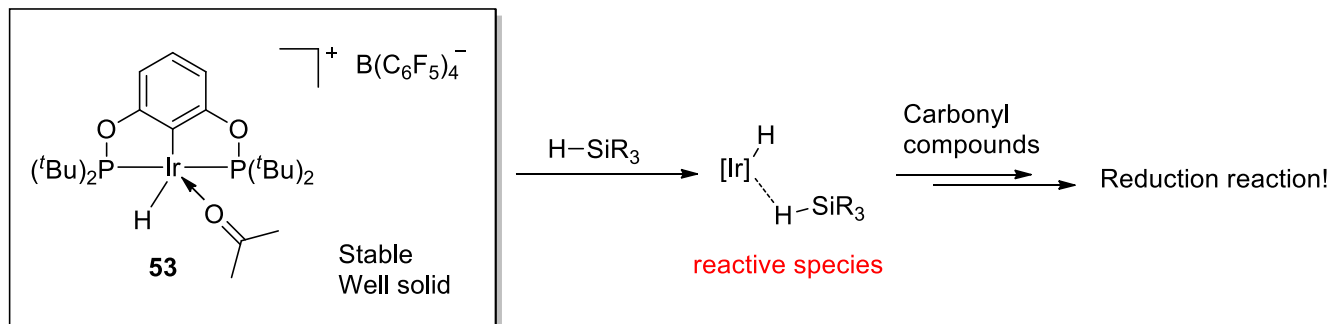
### 3. Reduction reaction

#### 7) Ir-Catalyzed Reduction of Carbon Dioxide to Methane.



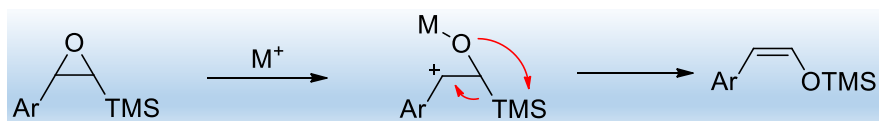
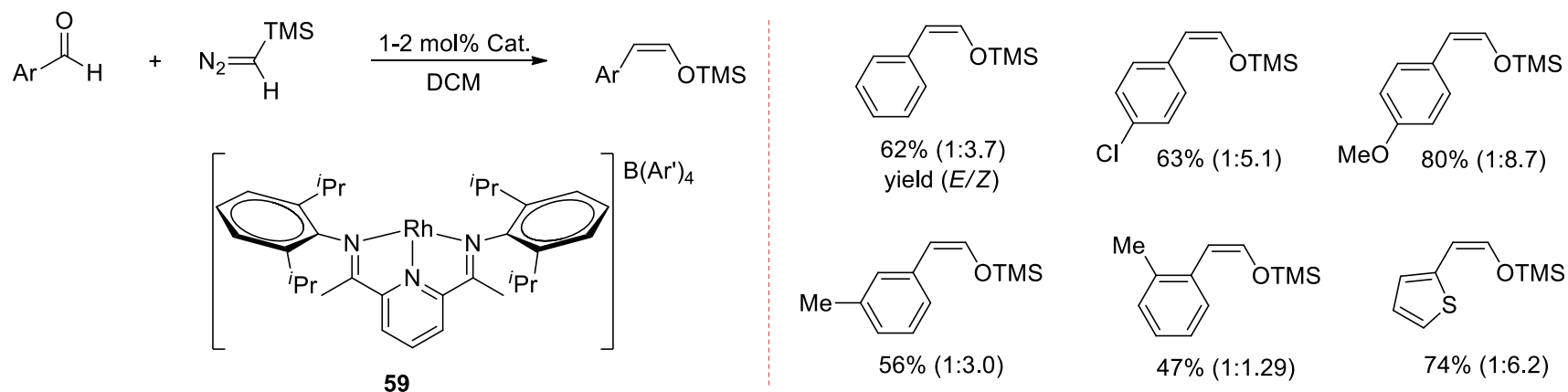
Park, S., et. al. *J. Am. Chem. Soc.* **2012**, *134*, 15708-15711.

#### 8) Summary for Ir Pincer complex catalyzed Reduction reactions.



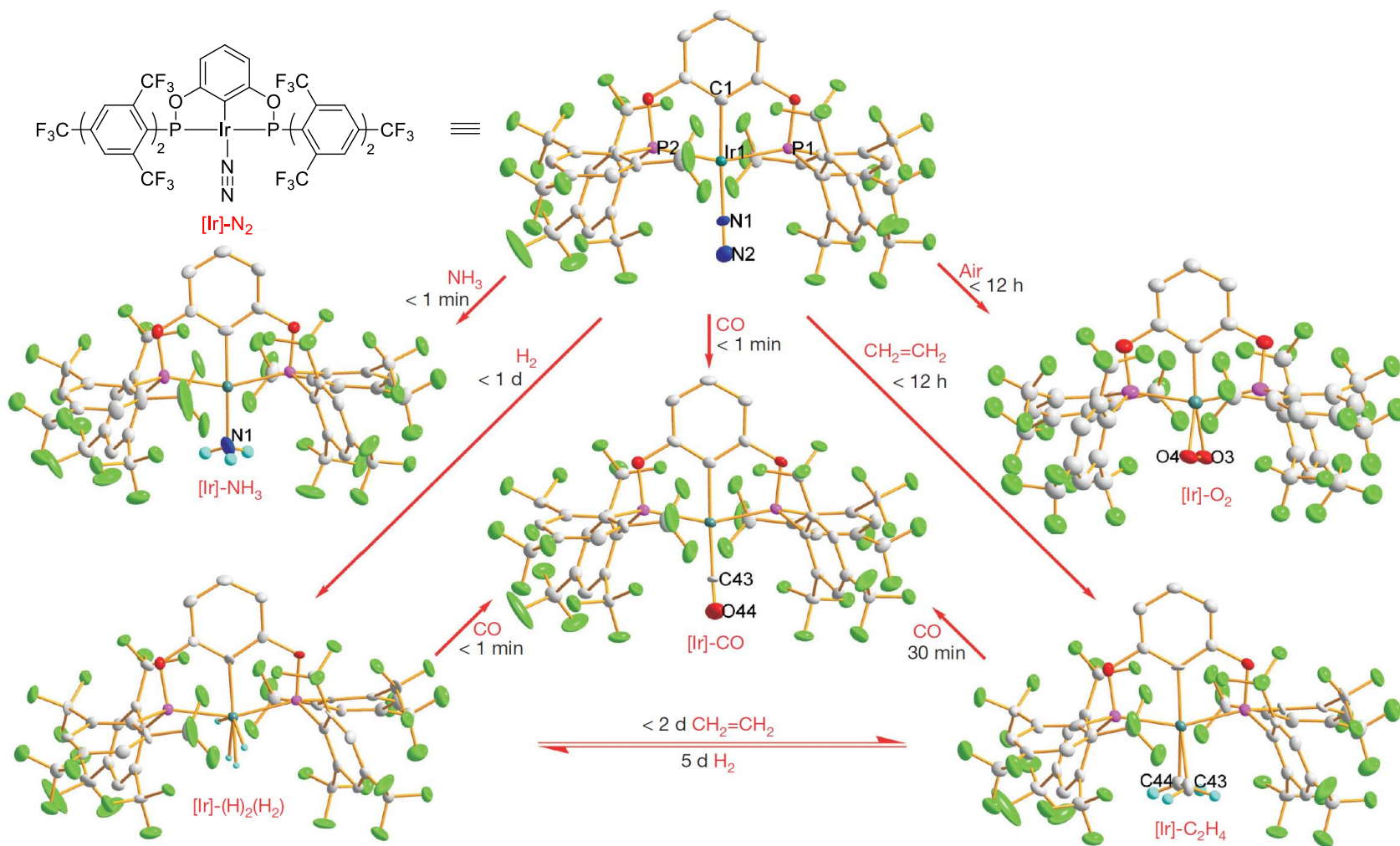
## 4. Miscellaneous

### 1) Rh (I)-Catalyzed Homologation of Aromatic Aldehydes with Trimethylsilyldiazomethane.



### 3. Miscellaneous

#### 2) Ligand exchange and Selective Catalytic Hydrogenation in Molecular Single Crystal.





# Acknowledgement

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